

```
> ##### GLOSS
#####
```

Gloss

if sulfur present assume as pyrite, in which case reduce O₂ by S₂/2, i.e., FeO -> FeS₂

~100 g

```
> H2O := 0.405;# 21.69 7.37
> CO2 := 0.068;# 3.67 3.04
> FeO := 0.073;# 3.89 5.27
> MgO := 0.062;# 3.30 2.51
> CaO := 0.106;# 5.69 6.02
> Na2O := 0.039;# 2.10 2.46
> K2O := 0.022;# 1.16 2.06
> Al2O3 := 0.117;# 6.26 12.04
> SiO2 := 0.975;# 52.25 59.23
> S2 := 0.0;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
    Al2O3/2 + SiO2 - S2/2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.00;
```

raw Gloss 0

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
H2O := 0.405
CO2 := 0.068
FeO := 0.073
MgO := 0.062
CaO := 0.106
Na2O := 0.039
K2O := 0.022
Al2O3 := 0.117
SiO2 := 0.975
S2 := 0.
O2 := 1.572000000
Al := 0.234
Na := 0.078
```

(1)

$$K := 0.044$$

$$C := 0.068$$

$$0.405, 0.068, 0.975, 0.234, 0.073, 0.062, 0.106, 0.078, 0.044, 1.572000000, 0.$$

```
> S2 := 0.00;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2/2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.00;
```

Gloss 0 + 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
   S2 := 0. (2)
```

$$O2 := 1.572000000$$

$$Al := 0.234$$

$$Na := 0.078$$

$$K := 0.044$$

$$C := 0.068$$

$$0.405, 0.068, 0.975, 0.234, 0.073, 0.062, 0.106, 0.078, 0.044, 1.572000000, 0.$$

```
> S2 := 0.01;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2/2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.00;
```

Gloss 0 + Sulfur + 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
   S2 := 0.01 (3)
```

$$O2 := 1.567000000$$

$$Al := 0.234$$

$$Na := 0.078$$

$$K := 0.044$$

$$C := 0.068$$

$$0.405, 0.068, 0.975, 0.234, 0.073, 0.062, 0.106, 0.078, 0.044, 1.567000000, 0.01$$

```
> ##### STAUDIGEL
#####
```

Staudigel

if sulfur present assume as pyrrhotite, in which case reduce O₂ by S₂, i.e., FeO -> FeS

~101.9 g

staudigel basalt 0.1528 0.069 0.7839 0.157 0.1434 0.167 0.2365 0.0345 0.0063 0.

```
> H2O := 0.1528;
> CO2 := 0.069;
> SiO2 := 0.7839;
> Al2O3 := 0.157 ;
> FeO := 0.1434 ;
> MgO := 0.167 ;
> CaO := 0.2365 ;
> Na2O := 0.0345 ;
> K2O := 0.0063;
> S2 := 0.0;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
    Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.00;
```

raw Stau

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
H2O := 0.1528
CO2 := 0.069
SiO2 := 0.7839
Al2O3 := 0.157
FeO := 0.1434
MgO := 0.167
CaO := 0.2365
Na2O := 0.0345
K2O := 0.0063
S2 := 0.
O2 := 1.458650000
Al := 0.314
```

$$Na := 0.0690$$

$$K := 0.0126$$

$$C := 0.069$$

$$0.1528, 0.069, 0.7839, 0.314, 0.1434, 0.167, 0.2365, 0.0690, 0.0126, 1.458650000, 0.$$

```
> S2 := 0. ;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.001;
```

Stau+ 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
   S2 := 0.                                                 (5)
   O2 := 1.458650000
   Al := 0.314
   Na := 0.0690
   K := 0.0126
   C := 0.070
0.1528, 0.070, 0.7839, 0.314, 0.1434, 0.167, 0.2365, 0.0690, 0.0126, 1.458650000, 0.
```

```
> S2 := 0.02;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.001;
```

Stau+ Sulfur + 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
   S2 := 0.02                                              (6)
   O2 := 1.438650000
   Al := 0.314
   Na := 0.0690
   K := 0.0126
   C := 0.070
```

0.1528, 0.070, 0.7839, 0.314, 0.1434, 0.167, 0.2365, 0.0690, 0.0126, 1.438650000, 0.02

> ##### BEHN
#####

BEHN

if sulfur present assume as pyrrhotite, in which case reduce O₂ by S₂, i.e., FeO -> FeS

~101.5 g (1.5 wt% h₂O)

```
> H2O := 0.083;
> CO2 := 0.;
> SiO2 := 0.903;
> Al2O3 := 0.140 ;
> FeO := 0.097 ;
> MgO := 0.304 ;
> CaO := 0.194 ;
> Na2O := 0.020 ;
> K2O := 0.001;
> S2 := 0.0;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
    Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.00;
```

raw behn + C + S₂

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
H2O := 0.083
CO2 := 0.
SiO2 := 0.903
Al2O3 := 0.140
FeO := 0.097
MgO := 0.304
CaO := 0.194
Na2O := 0.020
K2O := 0.001
S2 := 0.
O2 := 1.462500000
Al := 0.280
```

$$Na := 0.040$$

$$K := 0.002$$

$$C := 0.$$

$$0.083, 0., 0.903, 0.280, 0.097, 0.304, 0.194, 0.040, 0.002, 1.462500000, 0.$$

```
> S2 := 0.0;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.001;
```

behn + 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
```

$$S2 := 0.$$

(8)

$$O2 := 1.462500000$$

$$Al := 0.280$$

$$Na := 0.040$$

$$K := 0.002$$

$$C := 0.001$$

$$0.083, 0.001, 0.903, 0.280, 0.097, 0.304, 0.194, 0.040, 0.002, 1.462500000, 0.$$

```
> S2 := 0.02;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.001;
```

behn + Sulfur + 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
```

$$S2 := 0.02$$

(9)

$$O2 := 1.442500000$$

$$Al := 0.280$$

$$Na := 0.040$$

$$K := 0.002$$

$$C := 0.001$$

0.083, 0.001, 0.903, 0.280, 0.097, 0.304, 0.194, 0.040, 0.002, 1.442500000, 0.02

```
> ##### LOSIMAG
#####
```

LOSIMAG

if sulfur present assume as pyrrhotite, in which case reduce O₂ by S₂, i.e., FeO -> FeS

~95 g (4 wt% h₂O)

```
> H2O := .21090;
> CO2 := 0.0*H2O;
> SiO2 := 0.7075 ;
> Al2O3 := 0.037 ;
> FeO := 0.09711 ;
> MgO := 0.8666 ;
> CaO := 0.0527 ;
> Na2O := 0.0046 ;
> K2O := 0.00046;
> S2 := 0.0;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
    Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.00;
```

raw LOSIMAG

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
H2O := 0.21090
CO2 := 0.
SiO2 := 0.7075
Al2O3 := 0.037
FeO := 0.09711
MgO := 0.8666
CaO := 0.0527
Na2O := 0.0046
K2O := 0.00046
S2 := 0.
O2 := 1.379185000
```

(10)

```

        Al := 0.074
        Na := 0.0092
        K := 0.00092
        C := 0.

0.21090, 0., 0.7075, 0.074, 0.09711, 0.8666, 0.0527, 0.0092, 0.00092, 1.379185000, 0.

> S2 := 0.0;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.001;

```

LOSIMAG + 0.001 C

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
S2 := 0.                                (11)
```

$$O2 := 1.379185000$$

$$Al := 0.074$$

$$Na := 0.0092$$

$$K := 0.00092$$

$$C := 0.001$$

$$0.21090, 0.001, 0.7075, 0.074, 0.09711, 0.8666, 0.0527, 0.0092, 0.00092, 1.379185000, 0.$$

```

> S2 := 0.002;
> O2 := H2O/2 + CO2 + FeO/2 + MgO/2 + CaO/2 + Na2O/2 + K2O/2 + 3*
  Al2O3/2 + SiO2 - S2;
> Al := 2*Al2O3;
> Na := 2*Na2O;
> K := 2*K2O;
> C := CO2 + 0.001;

```

LOSIMAG + 0.001 C + S

```
> H2O,C,SiO2,Al,FeO,MgO,CaO,Na,K,O2,S2;
```

$$S2 := 0.002$$

$$O2 := 1.377185000$$

$$Al := 0.074$$

$$Na := 0.0092$$

$$K := 0.00092$$

(12)

$$C := 0.001$$

0.21090, 0.001, 0.7075, 0.074, 0.09711, 0.8666, 0.0527, 0.0092, 0.00092, 1.377185000, 0.002

```
> t0_deep := -2.059655E-10*z0^6 + 2.323113E-07*z0^5 - 1.076535E-04*
z0^4 + 2.625959E-02*z0^3 - 3.566382E+00*z0^2 + 2.582593E+02*z0 -
7.189476E+03; # z0 > 75.0 km
```

```
> t0_shallow := 1.255734E-06*z0^5 - 2.000554E-04*z0^4 +
1.180485E-02*z0^3 - 3.163565E-01*z0^2 + 6.026698E+00*z0 +
2.300000E+00 + .735544;
```

```
> z0 := p0/0.35;
```

$$\begin{aligned} t0_{\text{deep}} := & -2.059655 \cdot 10^{-10} z0^6 + 2.323113 \cdot 10^{-7} z0^5 - 0.0001076535 z0^4 + 0.02625959 z0^3 \\ & - 3.566382 z0^2 + 258.2593 z0 - 7189.476 \end{aligned} \quad (13)$$

$$\begin{aligned} t0_{\text{shallow}} := & 0.000001255734 z0^5 - 0.0002000554 z0^4 + 0.01180485 z0^3 - 0.3163565 z0^2 \\ & + 6.026698 z0 + 3.035544 \end{aligned}$$

$$z0 := 2.857142857 p0$$

```
> p := 10;
```

```
> subs(p0,t0_deep + 273);
```

```
> subs(p0=p,t0_shallow + 273);
```

$$p := 10$$

$$-1903.930548$$

$$355.9025519$$

```
> fsolve(t0_deep=t0_shallow,p0);
```

$$-1769.175064, -59.23176593, 26.17936336, 26.25000234 \quad (15)$$

```
> p0 := pbar/1000;
```

$$p0 := \frac{pbar}{1000} \quad (16)$$

```
> subs(pbar=26.1793e3,{t0_deep,t0_shallow});
```

$$\{301.97012, 301.9701589\} \quad (17)$$

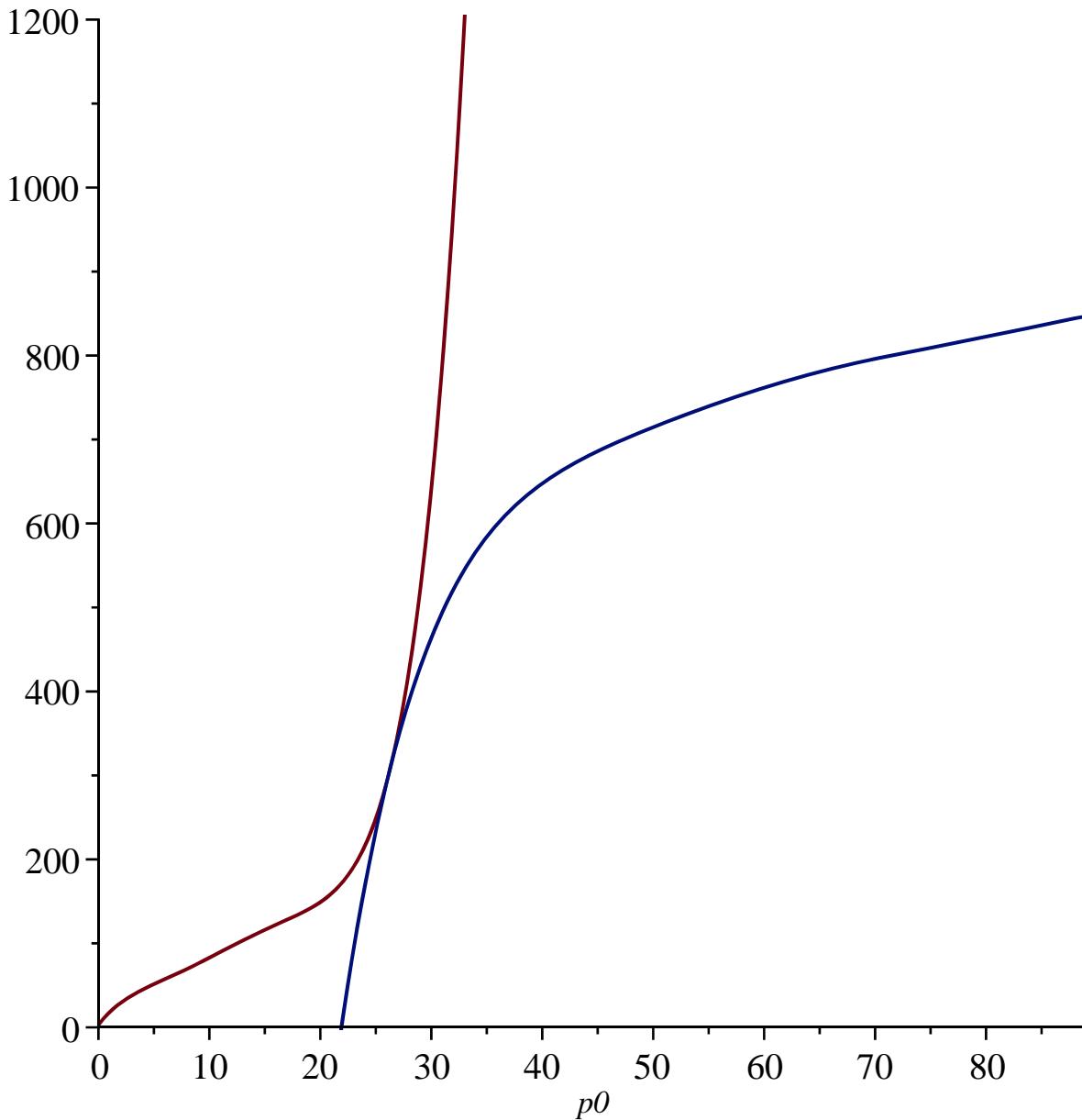
```
> t0_shallow;
```

$$\begin{aligned} 2.390878086 \cdot 10^{-19} pbar^5 - 1.333147188 \cdot 10^{-14} pbar^4 + 2.753317784 \cdot 10^{-10} pbar^3 \\ - 0.000002582502040 pbar^2 + 0.01721913714 pbar + 3.035544 \end{aligned} \quad (18)$$

```
> subs(pbar=A1,t0_deep);
```

$$\begin{aligned} -1.120433833 \cdot 10^{-25} A1^6 + 4.423134169 \cdot 10^{-20} A1^5 - 7.173910869 \cdot 10^{-15} A1^4 \\ + 6.124685714 \cdot 10^{-10} A1^3 - 0.00002911332244 A1^2 + 0.7378837142 A1 - 7189.476 \end{aligned} \quad (19)$$

```
> plot({t0_deep,t0_shallow},p0=0..255*.35,0..1200);
```



> $\text{eq1} := t1 = \text{dtdp} * p1 + \text{int};$ $\text{eq1} := t1 = \text{dtdp} p1 + \text{int}$ (20)

> $\text{eq2} := t2 = \text{dtdp} * p2 + \text{int};$ $\text{eq2} := t2 = \text{dtdp} p2 + \text{int}$ (21)

> $\text{solve}(\text{subs}(t1=1473, p1=2.5\text{e}4, t2=973, p2=4.5\text{e}4, \{\text{eq1}, \text{eq2}\}), \{\text{dtdp}, \text{int}\});$ $\{\text{dtdp} = -0.02500000000, \text{int} = 2098.\}$ (22)

> $1.353 * 18;$ 24.354 (23)

> $45000. * 0.05 + 223;$ 2473.00 (24)

```
[> c0 + var*(c1 + var*(c2 + var*(c3 + var*c4)))
```

$$\left[\begin{array}{l} > \text{exox} := 2*(1.585-1.572); \\ & \qquad \qquad \qquad exox := 0.026 \end{array} \right] \quad (25)$$

$$\left[\begin{array}{l} > \text{feox} := 2*\text{exox}; \\ & \qquad \qquad \qquad feox := 0.052 \end{array} \right] \quad (26)$$

$$\left[\begin{array}{l} > \text{fred} := 0.073 - \text{feox}; \\ & \qquad \qquad \qquad fred := 0.021 \end{array} \right] \quad (27)$$

$$\left[\begin{array}{l} > \text{feox}/\text{fred}; \\ & \qquad \qquad \qquad 2.476190476 \end{array} \right] \quad (28)$$