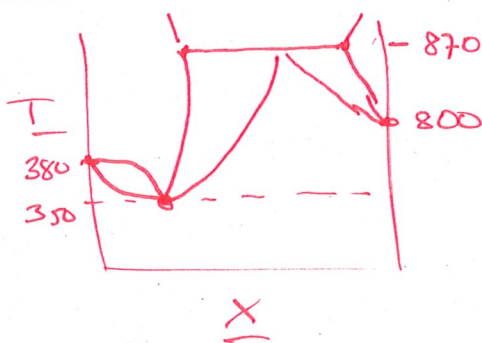


ans $g = \inf(f, x)$ assuming positive

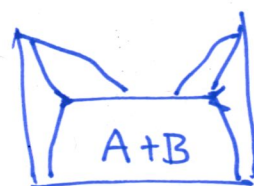
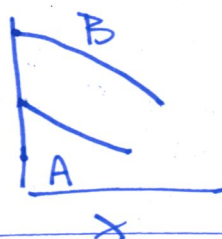
ans 1 $h = \text{diff}(f, x)$

```
plot (stats (a=1, b=2, c=3, {f,g,h}, x=-10..10, -20..20,
color = ["Red", ...
```

Example perfective



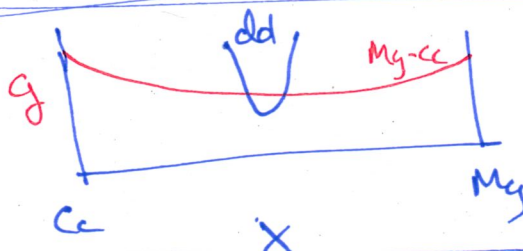
SKETCH! but not



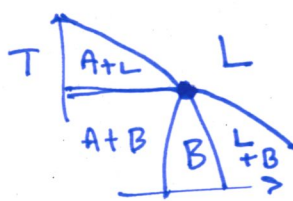
0.1 $\rightarrow$ Solvus



0.2 $\rightarrow$ order-disorder



0.3 $\rightarrow$



probable $\Rightarrow$

pentectic
(over)

collective
(old)

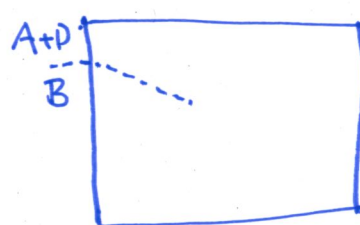
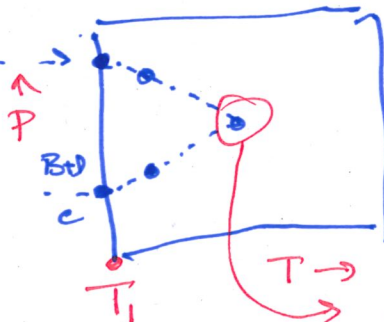
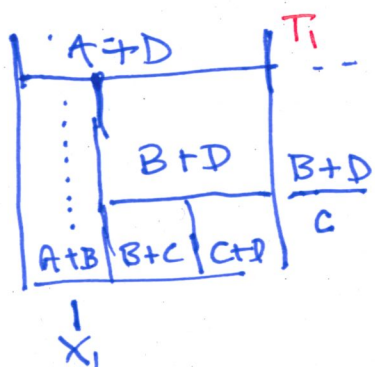


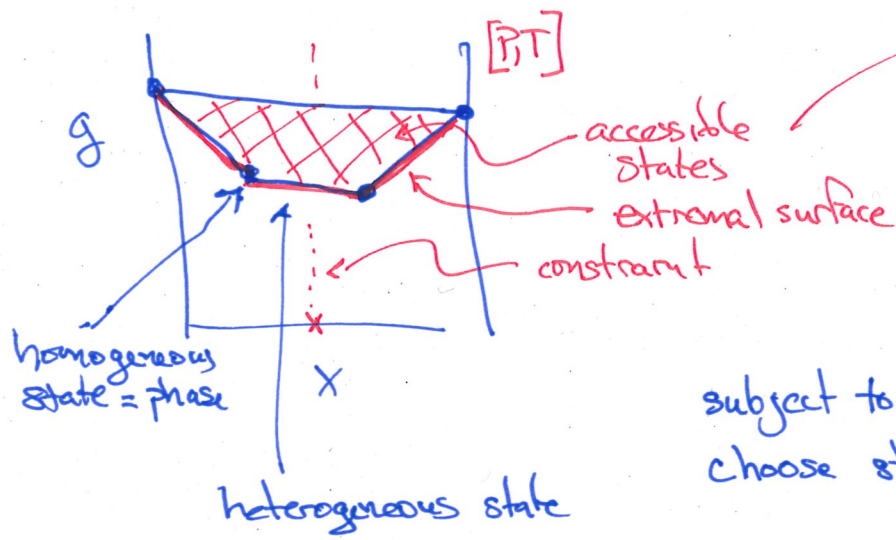
0.4



Projection

Section at x_1





subject to constraints x , $x \geq 0$
 choose stable state by an extremal principle $\rightarrow g$ is a min $\Rightarrow$

$$\frac{\partial g}{\partial x} = 0 \quad \frac{\partial^2 g}{\partial x^2} > 0$$

Thermodynamics $\rightarrow$ how thermal, mechanical, and chemical processes affect energy

Chemical thermo $\rightarrow$ no time $\rightarrow$ no kinetic energy; hydrostatic eg $\rightarrow$ rocks are fluid
 $\rightarrow$ scalar

4 laws

0th law - a legal issue $\rightarrow$ the transitive property of Eq.

$$T_A = T_B, T_B = T_C \Rightarrow T_A = T_C$$

Carathéodory 1910 (Born) $\rightarrow$ Fowler made it a law

1st Law

Clausius, 1850 $\rightarrow$ ETH

processed Q. heat
W work

$$dU - dQ_{\text{added}} + \sum dW_{\text{done}} = 0$$

internal energy

$$dU_{\text{universe}} = 0 \quad dU_{\text{sys}} \neq 0$$

state function

$$\oint dU = 0 \Rightarrow \text{exact differential}$$

state variables

$$M_1, \dots, M_k, V, T$$

$$\rightarrow \oint dM = 0$$

chemical state

of mass

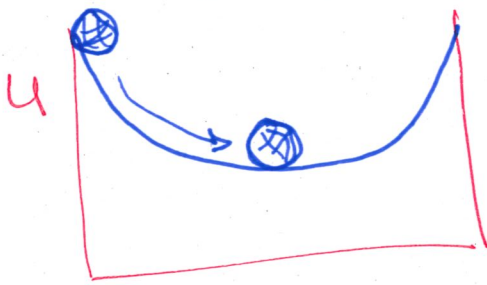
mechanical state

thermal state

Thermodynamic Work

2021 1.3

Mechanics



X

1st law $dU = -dW$

physics $dU = -F dx \rightarrow dW = F dx$

force $F = -\frac{dU}{dx}$

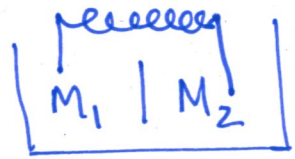
potential $\Theta = \frac{dU}{dx}$

$dU = \Theta dx$

Chemical Thermodynamics



dV
mechanical work



$dM_1, \dots, dM_2$
chemical work

1st law $dU = dQ - dW_{\text{mech}} - dW_{\text{chem}}$

$dW = -\frac{\partial U}{\partial V} dV = p dV$

$dU = dQ + \frac{\partial U}{\partial V} dV + \frac{\partial U}{\partial M} dM$

$dW_{\text{mech}} = p dV$ Force potential

$dW_{\text{chem}} = -\mu dM$

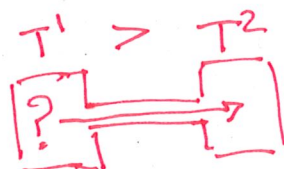
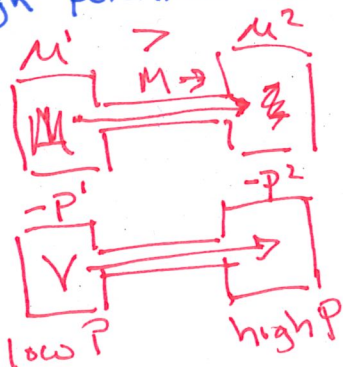
$dU = dQ - p dV + \mu dM$

$p(V, M, T) = -\frac{\partial U}{\partial V}$

$\mu(V, M, T) = \frac{\partial U}{\partial M}$

Equilibrium is a state where no processes are possible within the existing parts of a system $\Rightarrow$ potentials for every process must be uniform.

The property ψ_i is transferred from high potential Θ_i to low potential Θ_j



Conjugate variables

$\frac{\partial U}{\partial M_i} = \mu_i$ M_i & μ_i are conjugate

$\frac{\partial U}{\partial V} = -P$ V & P are conjugate

$\frac{\partial U}{\partial \psi_i} = \Theta_i$ ψ_i & Θ_i are conjugate

$dU = dQ$

$dU = T dX$

Thoughts on exactness

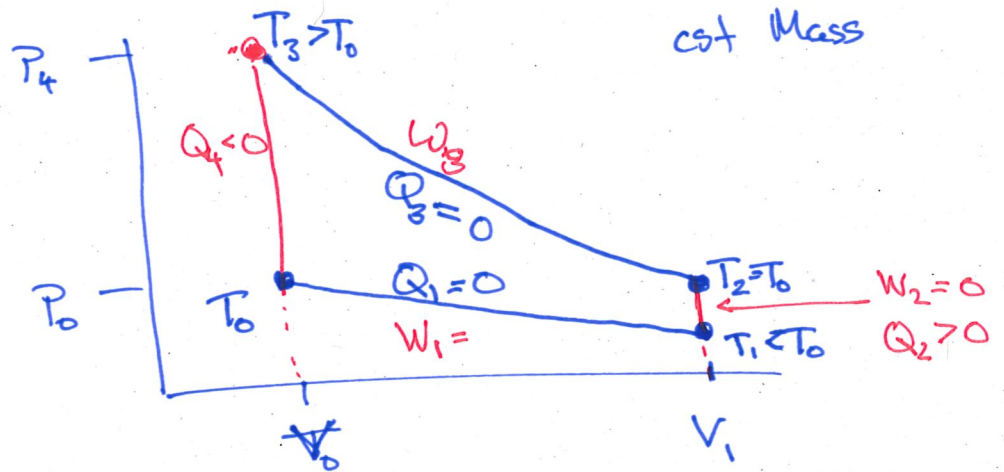
if $\delta Q = du + p dv - u dm$ and $\oint \delta T = 0$ how can it be exact?

$$PV = nRT$$

$$[PV^{5/8}]_Q = \text{const}$$

$$W = \int p dv$$

$$P = P_0 \left[\frac{V_0}{V} \right]^{5/8}$$



$$\oint \delta u = 0 = \sum Q - \sum W$$

$$|W_3| > |W_1| \Rightarrow \sum W < 0$$

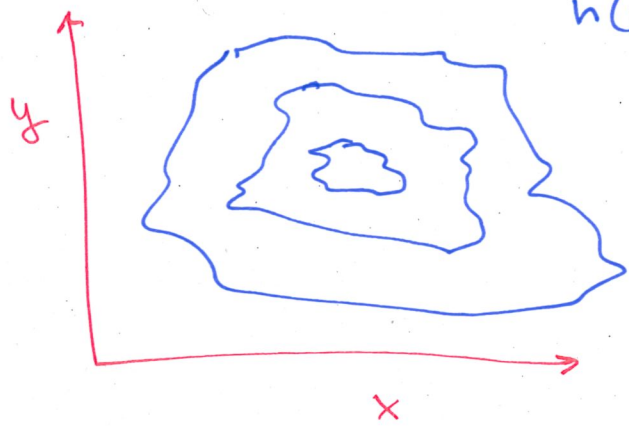
$$\Rightarrow \sum Q > 0$$

problem 1.2

$$W1 := \int_{V_0}^{V_1} P_0 \left(\frac{V_0}{V} \right)^{5/8} dV$$

... assuming positive;

or assuming $V_0 > 0, V_1 > 0$;



$$dh = dW_x + dW_y$$

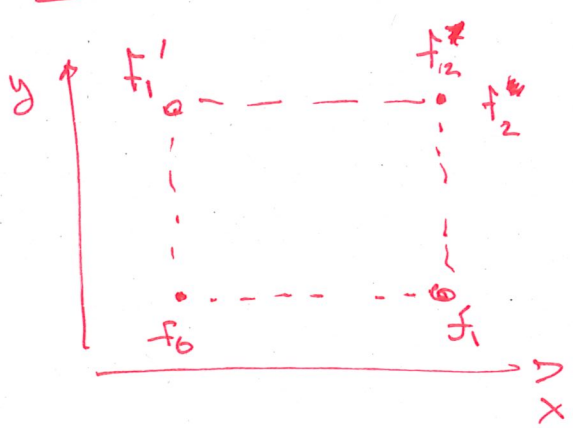
$$dh = \frac{\partial h}{\partial x} dx + \frac{\partial h}{\partial y} dy$$

$$= p(x,y) dx + q(x,y) dy$$

$$dU_Q = (-p) dV + \mu dM$$

$$U_Q(V, M)$$

What if you can't see The mountain?



Taylor Series
$$f(x) = \sum_{n=0}^{\infty} \frac{f^{(n)}(x_0)}{n!} (x-x_0)^n$$

$n=1$
$$f(x) = f(x_0) + \left. \frac{\partial f}{\partial x} \right|_{x=x_0} \delta x$$

$\Rightarrow$

$$f_0 \rightarrow f_1 \rightarrow f_2$$

$$f_1 = f_0 + \left. \frac{\partial f}{\partial x} \right|_{0,0} \delta x$$

$$f_2 = f_1 + \left. \frac{\partial f}{\partial y} \right|_{\delta x, 0} \delta y$$

$$f_2 = f_0 + \left. \frac{\partial f}{\partial x} \right|_{0,0} \delta x + \left. \frac{\partial f}{\partial y} \right|_{\delta x, 0} \delta y$$

$$f_2' = f_0 + \left. \frac{\partial f}{\partial y} \right|_{0,0} \delta y + \left. \frac{\partial f}{\partial x} \right|_{0,\delta y} \delta x$$

If $f_2 = f_2'$ then

$$\left. \frac{\partial f}{\partial x} \right|_{0,0} \delta x + \left. \frac{\partial f}{\partial y} \right|_{0,\delta y} \delta y = \left. \frac{\partial f}{\partial y} \right|_{0,0} \delta y + \left. \frac{\partial f}{\partial x} \right|_{\delta x,\delta y} \delta x$$

$$\left. \frac{\partial f}{\partial x} \right|_{0,\delta y} \delta x - \left. \frac{\partial f}{\partial x} \right|_{0,0} \delta x = \left. \frac{\partial f}{\partial y} \right|_{\delta x,\delta y} \delta y - \left. \frac{\partial f}{\partial y} \right|_{0,0} \delta y$$

$$\frac{\left(\left. \frac{\partial f}{\partial x} \right|_{0,\delta y} - \left. \frac{\partial f}{\partial x} \right|_{0,0} \right)}{\delta y} = \frac{\left(\left. \frac{\partial f}{\partial y} \right|_{\delta x,\delta y} - \left. \frac{\partial f}{\partial y} \right|_{0,0} \right)}{\delta x}$$

infinitesimal limit

$$\frac{\partial^2 f}{\partial x \partial y} = \frac{\partial^2 f}{\partial y \partial x}$$

Euler's criterion for exactness $\Rightarrow$ or better

$$\frac{\partial}{\partial x} \left(\frac{\partial f}{\partial y} \right) = \frac{\partial}{\partial y} \left(\frac{\partial f}{\partial x} \right)$$

Euler 1767-1787
10 CHF e, z, f, etc
Catherine the great

2021 1.6

$$dU_Q = (-p)dV + \mu dM$$

$$\frac{\partial}{\partial V} \left(\frac{\partial U}{\partial M} \right) = \frac{\partial}{\partial M} \left(\frac{\partial U}{\partial V} \right)$$

$$\frac{\partial \mu}{\partial V} = \frac{\partial (-p)}{\partial M}$$

Problem 1.3 - don't use maple

No set up script for chapter 1 problems

Problem 1.4 will be covered next week. The problem set is
NOT assigned yet.